Year 10 TRIPLE - Chemistry Name:		
Sept - Oct	 lons / ionic bonding / properties Covalent bonding Molecular compounds Allotropes of carbon – diamond / graphite / fullerenes / graphene Metallic bonding Bonding models (ionic, covalent, metallic) Acids and indicators Concentration of acids 	
Nov - Dec	 Bases and salts	
Jan - Feb	 Extraction of metals – carbon / electrolysis / biological Corrosion and oxidation Recycling and lifecycle assessments Electrolysis of copper sulphate Core practical 4 Products of electrolysis Transition metals Corrosion Sacrificial protection Electroplating 	
Feb - March	 Alloying Reversible reactions Fertilisers Haber process Factors affecting equilibrium % Yield and Atom Economy 	
April - May	 Concentration calculations Titrations Molar gas volumes Chemical and fuel cells Group 1 – Alkali metals 	
June – July	 Group 7 – Halogens Group 0 – Noble gases Exo and Endothermic Reactions Reaction Profiles Bond Energy Calculations 	

Test 1 – (Bonding) Test 2 – (Acids and calculations) Test 3 (Electrolysis, metals, transition metals and alloys) Test 4 – _ Reversible Reactions Test 5 Quantitative analysis and Fuel Cells

YEAR 10 EXAM

Year 11	TRIPLE - Chemistry Name:
Sept - Oct	 Rates of reaction Factors affecting rates of reaction Core Practical 6 Catalysts and activation energy Exothermic and endothermic reactions Reaction profiles Bond energy calculations
Nov - Dec	 Crude oil Fractional distillation Alkanes Combustion of fuels Cracking Early and modern atmosphere
Jan - Feb	 Climate change Alkanes and alkenes Ethanol production Alcohols Core Practical 8
Feb - March	 Flame tests Core Practical 7a Identifying ions Core Practical 7b & c Reactions of carboxylic acids Polymers
April - May	Choosing materials
Test 1 Groups and Energy Changes Test 2 Quantitative Chemistry & Chemical Cells Test 3 Fuels and Earth & Atmosphere Test 4 Organic Chemistry MOCK 1 MOCK 2	